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been at the forefront of the development agenda
yet women worldwide only enjoy an average of 77
percent of the legal rights

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web the ions which do not participate in chemical
reactions and present the same on both sides of the
reactions are called spectator ions in the net
chemical reaction the spectator ions are cancelled
from both sides of the equation in other words
spectate means watch the other ions react in an
aqueous solution

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web these ions are called spectator ions because they
do not participate in the actual reaction canceling
the spectator ions gives the net ionic equation
which shows only those species that participate in
the chemical reaction 2ag aq cr 2o 7 2 aq →
ag 2cr 2o 7 s label 4 2 3

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web chemical equation in which all dissolved ionic
reactants and products including spectator ions are
explicitly represented by formulas for their
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web jan 29 2020 the ionic form of this reaction is 2
na aq 2 cl aq cu 2 aq so 4²⁻ aq 2 na aq so 4²⁻ aq cucl 2 s
the sodium ions and sulfate ion are the spectator ions
in this reaction they appear unchanged in both the
product and reactant side of the equation

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na aq hco 3 aq na aq cl aq h₂o l co₂ g the spectator
ions are cl and na net ionic equation h aq cl aq na aq
hco 3 aq na aq cl aq h₂o l co₂ g h aq hco 3 aq h₂o l
co₂ g here is a video on spectator ions

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web the six no 3 aq ions and the six na aq ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water h aq oh aq h 2 o l the spectator ions in koh aq and hno 3 aq are k aq and no 3 aq

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both before and after the reaction the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction

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explanation in the net ionic equations we are not include the spectator ions in the equations spectator ions the ions present on reactant and product side 16 used to grind chemicals to powder tmraor nda stlepe star 4 4 5 heart 181 verified verified answer food

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web jun 2 2023 the spectator ions are cut out from a reaction hence the net ionic equation for the above

reaction would be $2\text{Cl}^-\text{aq} + \text{Cu}^{2+}\text{aq} \rightarrow \text{CuCl}_2\text{s}$ these spectator ions are not actively involved in the reaction the existence of spectator ions affects the debye length

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