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her final displacement

from the starting point

verified answer

engineering the shaft

rotating at 480 mathrm

rpm 480 rpm carries a 10

text in 10 in diameter

chain sprocket at c c

that receives 11 mathrm

hp 11 hp from a mating

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the left as shown

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web produce saliva and

amylase and begin the

process of chemical

digestion gastric glands

line the stomach and

contain specialized

cells that secrete

hydrochloric acid

protective mucus and

produce pepsin pancreas

endocrine exocrine

functions endocrine

ductless secrete

hormones that travel to

their target through the

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starts with the muscular

system then moves to the

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finally students examine

the main organs of the

body cavity which is

mostly the digestive

system

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system testes ovary

uterine horn vagina

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kidney digestive system

greater omentum

mesentery stomach liver

esophagus gallbladder

pancreas spleen bile

duct cecum colon rectum

small intestine duodenum

jejunum ileum

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list of keys 90 12 2 80
55 11 17 15 58 98 36
this list is to be
sorted using the
insertion sort algorithm
as described in this
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lists show the resulting
list after seven passes
of the sorting phase
that is after seven
iterations of

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lungs on a paper towel
and find the superior
inferior and middle
lobes sketch and label
below 16 cut the heart
in half and observe the
chambers of the heart
sketch and label the
ventricle and atria 17
if your cat is female
locate the uterine horns
the ovaries and the
vagina

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system function organs
anatomy cleveland clinic
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organs that make up the
digestive system in
order of their function
are the mouth esophagus
stomach small intestine
large intestine rectum
and anus helping them
along the way are the
pancreas gall bladder
and

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of nitrogen at 748 mm hg
are compressed to 725 mm
hg at constant
temperature what is the
new volume 2 a gas with
a volume of 4 0l at a
pressure of 205kpa is
allowed to expand to a
volume of 12 0l

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example 1 1 a 17 50ml

sample of gas is at 4
500 atm what will be the
volume if the pressure
becomes 1 500 atm with a
fixed amount of gas and
temperature solution v2
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laws 4 v 2 p 1 v 1 p 2 4
500atm 17 50ml 1 500atm
gas laws 5 gas laws 5 4
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500 a t m

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relationship into one
equation each of these
laws can be derived from
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law pv t k v1p1t2 v2p2t1
p 1 v 1 t 1 p 2 v 2 t 2
p t k p1t2 p2t1 p 1 t 1
p 2 t 2 v t k v 1 t 2 v
2 t 1 1 1 boyle s law
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a volume of 17 liters

has a pressure of 2 3
atm and a temperature of
299 k if i raise the
temperature to 350 k and
lower the pressure to 1
5 atm what is the new
volume of the gas 2

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to calculate the value
of an unknown that
corresponds to one of
the principal measurable
quantities of gases

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the choice that best
completes the statement
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depth a volume b surface
area 2 why does a can
collapse when a vacuum

pump removes air from the can a

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what are the gas laws
the gas laws are a group
of laws that govern the
behaviour of gases by
providing relationships
between the following
the volume occupied by
the gas the pressure
exerted by a gas on the
walls of its container
the absolute temperature
of the gas

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p1v1 p2v2 1 atm 760 0 mm
hg 101 3 kpa if 22 5 l
of nitrogen at 748 mm hg
are compressed to 725 mm
hg at constant

temperature what is the
new volume a gas with a
volume of 4 0l at a
pressure of 205kpa is
allowed to expand to a
volume of 12 0l

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boyle s law t and n
constant charles law p
and n constant combined
gas law n constant p1 v1
p2 v2 v v ideal gas
equation pv nrt ideal
gas equation pv nrt

ideal gas equation
example 2 31 g of some
gas has the volume of 0
706 l at 2 5 atm and
room temperature 25 c
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gay lussac s law boyle s
law charles law combined
gas law ideal gas law b
state the following laws
charles law the volume v
of a given mass of a gas
at constant pressure p
is directly proportional
to its temperature t
answers to gas laws
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