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the reactant that a is used up at the  
end of the reaction b is left over  
after the reaction c has the largest  
number of atoms d has the fewest  
number of atoms answer correct answer  
a is used up at the end of the  
reaction question

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reactants excess reactants and  
theoretical yield in a chemical  
reaction select one of two different  
reactions choose the number of  
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observe the products created and the  
reactants left over

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becomes  $\text{H}_2\text{O}$  is selected

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atoms are needed correct answer c  
aluminum is limiting 10 more atoms  
are needed according to the following  
balanced equation 2 formula units of  
iron iii oxide  $\text{Fe}_2\text{O}_3$  can be formed by  
reacting 4 atoms of iron  $\text{Fe}$  with 3  
molecules of oxygen gas  $\text{O}_2$

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reactants are shown on the left and  
products are shown on the right o for  
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describes the reaction of sodium  
Na and chlorine gas  $\text{Cl}_2$  to form  
table salt

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