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university q1
given the
following
reaction h_2so_4
 naoh na_2so_4 h_2o
 h_2s o_4 n a o

h_2n a 2 s o_4 h
2 o if it takes
27 4 ml of 0
768 m naoh n a
o h to titrate
16 7 ml of
 h_2so_4 h_2s o_4
what is the
concentration
of the h_2so_4 h_2s o_4
solution hint
balance the
equation first
q2
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problem sheet 1
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chemistry
stoichiometry
problem sheet 1
directions
solve each of
the following
problems show
your work
including
proper units to
earn full
credit 1 silver
and nitric acid
react according
to the

following
balanced
equation 3ag s
 $4\text{hno}_3\text{aq}$ 3
 agno_3aq 2h_2
 o l no g a
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are pV and $n V_m$
rt where V_m 22
4 dm³ mol⁻¹ at

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web 1 carbon
monoxide reacts
with oxygen to
produce carbon
dioxide answer
the following
questions for

the reaction of
1 0 lof carbon
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oxygen at stp a
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equation 20d or
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are required to
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silver nitrate
reacts with

barium chloride
to form silver
chloride and
barium nitrate
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practice

problems using
the following
equation 2 nh_3
 $\text{g} + \text{h}_2 \text{ so}_4 \text{ nh}_4$
 $2 \text{ so}_4 \text{ aq}$

calculate the
volume of nh_3
in liters

needed at 20°C

and 25 atm to

react with 150

kg of $\text{h}_2 \text{ so}_4$

5 2 1 practice

problems

reaction

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if gasoline

contains 84 2

carbon by mass

and has a

density of 0

8205 g ml

determine the

mass of carbon

dioxide

produced during

a 500 mile trip

3 785 liters

per gallon

answer 1 28 10

5 g co_2

gas

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worksheet**

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worksheet

please answer

the following

on separate

paper using

proper units

and showing all

work please

note that these

problems

require a

balanced

chemical

equation 1

carbon monoxide

reacts with

oxygen to

produce carbon

dioxide if 1 0

l of carbon

monoxide reacts

with oxygen at

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gas stoich

answers pdf

mole unit

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your problem	2 2 c 6 h 10 17	<u>assessment test</u>
solving	o 2 12 co 2	<u>i want to be a</u>
approach using	using the	<u>ballet dancer</u>
internationally	following	<u>i need help</u>
accepted	equation to	<u>with algebra</u>
symbols for	answer	<u>i am blind and</u>
elements	questions 8 11	<u>my dog is dead</u>
quantities	2 c 6 h 10 17 o	<u>i hate my best</u>
numbers and	2 12 co 2 10 h	<u>friend</u>
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