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normally atoms are

neutral they have
the same number of
protons in the
nucleus as they
have electrons
orbiting in the
energy levels
around the nucleus
atoms

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ions and molecules
the building blocks*

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isotopes and ions

isotopes isotopes of

an element have

the same number of

protons but

different numbers

of neutrons in their

nuclei isotopes of

an element have

atomic number

mass number and

isotopes video khan

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atoms are neutral

which means there

is an equal number

of protons and

electrons ions are

atoms with a

charge the charge
can be positive or
negative in

2 atoms molecules

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web some isotopes
may emit neutrons
protons and
electrons and attain
a more stable
atomic

configuration lower

level of potential

energy these are

radioactive isotopes

or

2 2 atomic

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2 6 protons

neutrons and electrons in

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