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Additions to C-X ?-Bonds Barry M. Trost,1992-09-08 Volume 2 deals mainly with the addition reactions of delocalized carbanions (enolates) and their synthetic relatives (metalloenamines, enol ethers, allyl organometallics) with carbonyl compounds, imines and iminium ions. Major emphasis is placed on C-C bond-forming reactions such as the aldol and Mannich reactions. Acylation reactions are also included in this volume. Several topics that have not previously been reviewed are covered, including the use of enzymatic aldol reactions in synthesis and the Passerini-Ugi reactions.

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g **lab 334 nitration of aromatic compounds preparation of studocu** - Dec 23 2022

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic

source in the nitration of bromobenzene comes from a nitronium ion no2 which acts as the electrophilic species

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