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web answer based on the data the strong acid strong base combination of

hydrochloric acid and sodium hydroxide had the neutral pH at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point [pdf acid base titration researchgate](#) - Nov 27 2022 web oct 1 2019 abstract titration is an analytical method used in biomedical sciences and analytical chemistry laboratories to determine the quantity or the concentration of a known or unknown substance there **acid base titration lab**

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web 1 approximately 3 g of hydrated oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ was weighted to the nearest 0.1 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of hcl because this is a

strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the pH of the solution is then pH log 0.100 = 1.000 pH log 0.100 = 1.000 b titrant volume 12.50 ml

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web one type of titration uses a neutralization reaction in which an acid and a base react to produce

a salt and water in equation 1 the acid is HCl hydrochloric acid and the base is NaOH sodium hydroxide when the acid and base react they form NaCl sodium chloride which is also known as table salt

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